Synthesis and transformations of metallacycles 30.* Aluminacyclopentanes in the synthesis of secondary and tertiary alcohols

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A method for the synthesis of secondary and tertiary alcohols was developed. The method is based on cycloalumination of olefins with trialkylalanes in the presence of Zr-containing catalysts followed by the CuCl-catalyzed in situ reaction of the corresponding aluminacyclopentanes with aldehydes or ketones.

Key words: olefins, cycloalumination, trialkylalanes, aluminacyclopentanes, catalysis by CuCl, aldehydes, ketones, secondary and tertiary alcohols.

According to the published data, ^{2–4} cycloalumination of olefins with trialkyl- or alkylhaloalanes in the presence of Zr-containing catalysts (Cp₂ZrCl₂, ZrCl₄) affords substituted aluminacyclopentanes (ACP) in high yields (80-95%).

The synthetic value of the proposed reaction substantially increased after the resulting ACP had been found to be easily convertible into practically important thiophanes,⁵ selenophanes,⁶ cyclopropanes,⁷ cyclobutanes, 8,9 2-vinylalkanes, 10,11 and 2- or 1,2-substituted 1.3-dienes.³

In this work, with the aim of developing a new method for the synthesis of secondary and tertiary alcohols from olefins and trialkylalanes, in situ catalytic reactions of ACP 1 with aldehydes and ketones 2 in the presence of transition metal salts and complexes were studied.

Results and Discussion

It was found that hydrolysis of the reaction mixture obtained by the reaction of 1-ethyl-3-(n-hexyl)aluminacyclopentane¹² with butyraldehyde in the presence of 10 mol.% CuCl at 20-22 °C affords a mixture of four secondary alcohols. These are diastereomeric 7-methyltridecan-4-ols (3a) and 6-ethyldodecan-4-ols (4a) in a ratio of 55:45 (GLC data); the total yield was ~80% (Scheme 1). In both cases, the ratio between diastereomers 3a and 4a is ~1:1, which follows from the nearly equal integral intensities of the corresponding separate signals in the ¹³C NMR spectrum.

Scheme 1

$$R^{1} + Et_{3}AI \xrightarrow{[Zr]} -C_{2}H_{6}$$

$$+ Et_{3}AI \xrightarrow{[Zr]} -C_{2}H_{6}$$

$$+ Et_{3}AI \xrightarrow{[Zr]} -C_{2}H_{6}$$

$$+ Et_{3}AI \xrightarrow{[Zr]} -C_{2}H_{6}$$

$$+ R^{1} \xrightarrow{R^{2} - C - R^{3}} \xrightarrow{2} -C_{2}H_{3}$$

$$+ R^{2} \xrightarrow{R^{3}} -R^{2} \xrightarrow{R^{3}} -R^{2} \xrightarrow{R^{3}} -R^{3}$$

$$+ R^{2} \xrightarrow{R^{3}} -R^{3} \xrightarrow{R^{2} - R^{3}} -R^{2} \xrightarrow{R^{3}} -R^{3}$$

$$+ R^{4} \xrightarrow{R^{2} - C - R^{3}} \xrightarrow{R^$$

In the above reaction, CuCl is most active among the studied complexes and salts of transition metals and rareearth elements (Cu, Mn, Fe, Co, Ni, Pd, Zr, and Rh), which agrees with the data on addition of zirconacyclopentanes to a carbonyl group in the presence of copper salts. 13 For instance, the use of PdCl₂ or NiCl₂ instead of

^{*} For Part 29, see Ref. 1.

Scheme 2

5, **7**: Δ^4 (**a**), Δ^3 (**b**); **8**: $R^1 = Pr^n$, $R^2 = H$; **9**: $R^1 = Ph$, $R^2 = H$; n = 1—3

CuCl reduces the total yield of regioisomers **3a** and **4a** to 61 and 40%, respectively. The reaction did not occur at all without a catalyst.

The synthesis of secondary alcohols $\bf 3a$ and $\bf 4a$ involving the intermediacy of ACP is of general character. For example, $\bf 3$ -(n-alkyl)-1-ethylaluminacyclopentanes react with aldehydes containing alkyl, isoalkyl, and aryl substituents to give alcohols $\bf 3b$ — $\bf d$ and $\bf 4b$ — $\bf d$ in $\bf 75$ — $\bf 85\%$ yields (see Scheme 1). The $\bf R^2COR^3$: ACP ratio was $\bf 12:10$ (10 mol.% CuCl).

It was found that ACP are also very reactive toward structurally various ketones, giving the corresponding tertiary alcohols. For instance, the reaction of alkyl and phenyl ketones with 1-ethyl-3-(n-hexyl)aluminacyclopentane in the presence of 10 mol.% CuCl (20–22 °C, 8 h) affords mixtures of tertiary alcohols **3e,f** and **4e,f** in the ~3: 2 ratio; the total yield was ~80% (see Scheme 1).

Addition to a carbonyl group predominantly involves the Al—C bond, which is most distant from the alkyl substituent in ACP. This suggests that the reaction regioselectivity can be significantly enhanced when ACP contains bulky substituents at one of the Al-bonded carbon atoms. This assumption was experimentally confirmed

Scheme 3

$$\begin{pmatrix}
A_{1} & B_{1} & B_{2} & B_{3} & B_{4} & B$$

 $R = Pr^{n}(\mathbf{a}), Pr^{i}(\mathbf{b})$

by the reaction of 2,3-disubstituted ACP 5 ¹⁴ and 6 ¹⁵ with acetone, butyraldehyde, and benzaldehyde (Scheme 2).

Aluminacyclopentanes bearing symmetrically arranged substituents react with aldehydes. For instance, the reaction of *trans*-3,4-di(*n*-butyl)aluminacyclopentane 10 ¹⁶ with butyraldehyde or isobutyraldehyde affords a mixture of diastereomeric secondary alcohols 11a or 11b, respectively (Scheme 3).

Thus, starting from olefins and aldehydes or ketones, one can obtain various secondary and tertiary alcohols through the *in situ* formation of ACP.

Experimental

The catalyst Cp₂ZrCl₂ was prepared according to a known procedure. 17 Commercial CuCl was used. Solvents were dried immediately before use (over LiAlH₄ for hexane and over Na metal for toluene) and distilled. Reactions with organometallic compounds were carried out under dry argon. Commercially available 95% Et₃Al and 88.6% Bui₃Al (OAO "Redkino pilot plant") were used. Hydrolysis products were analyzed by GLC on a Chrom-5 chromatograph in a helium flow (column 1200×3 mm, 5% SE-30 or 15% PEG-6000 on Chromaton N-AW). HPLC analysis of fullerene-containing alcohols was performed on a Hewlett Packard instrument (model 1050) (column 250×4 mm, Silasorb-600 (8 µm), toluene as the mobile phase; UV detection was at $\lambda = 333$ nm). IR spectra were recorded on a UR75 spectrometer (thin film). Mass spectra were recorded on an MX-1306 spectrometer (70 eV, 200 °C). ¹H and ¹³C NMR spectra were recorded on JEOL FX-90 Q (90 (1H) and 22.5 MHz (13C)) and Bruker AM-300 instruments (300.13 (¹H) and 75 MHz (¹³C)) in CDCl₃; the spectra were edited with the JMOD procedure. Chemical shifts are given on the δ scale with Me₄Si as the internal standard. The yields of organoaluminum compounds were determined by GLC analysis of hydrolysis products. The mixtures of regioisomers (3 and 4) were separated on a Carlo Erba Fractovap Mod. GW preparative gas chromatograph in a helium flow (column 4000×6 mm, stationary phase SE-30). Compounds 7 and 11 were isolated by column chromatography on silica gel L (100-250 µm) in hexane—Et₂O (4:1). Compounds 8 and 9 were chromatographed on Al₂O₃ (successive elution with toluene and methanol).

CuCl-catalyzed reactions of 3-alkyl-1-ethyl-ACP with aldehydes and ketones (general procedure). The catalyst Cp₂ZrCl₂ (0.5 mmol), hexane (3 mL), an olefin (10 mmol), and Et₃Al (12 mmol) were placed in a glass reaction vessel under dry argon at 0 °C. The solution was stirred at ~20 °C for 12 h. Hexane (3 mL) and, at $-15 \,^{\circ}\text{C}$, CuCl (1 mmol) were added. Then an aldehyde or ketone (12 mmol) was slowly added dropwise. The reaction mixture was stirred at ~20 °C for 8 h and treated with aqueous 8-10% HCl or 15% DCl in D₂O. Products 3 and 4 were extracted with ether or hexane.

7-Methyltridecan-4-ols* (3a). The yield was 44%, $R_{\rm f}$ 0.48. Found (%): C, 78.02; H, 13.86. C₁₄H₃₀O. Calculated (%): C, 78.43; H, 14.10. IR, v/cm⁻¹: 3380, 2945, 2900, 2850, 1100, 1255, 750. ¹H NMR, δ: 0.84–0.93 (m, 9 H, Me); 1.26–1.63 (m, 19 H, CH₂, CH); 3.45-3.68 (m, CH-OH). ¹³C NMR, δ : 14.06 (q, C(13)); 14.09 (q, C(1)); 18.80, 18.84 (t, C(2)); 19.60, 19.72 (q, C(14)); 22.66 (t, C(12)); 26.98, 27.02 (t, C(9)); 29.65 (t, C(10)); 31.92 (t, C(11)); 32.81 (t, C(6)); 32.90, 32.95 (d, C(7)); 34.95, 35.01 (t, C(5)); 36.92, 37.10 (t, C(8)); 39.63, 39.71 (t, C(3)); 72.04, 72.16 (d, C(4)). MS, m/z: 196 [M – 18]⁺.

4-Methyl-1-phenyldecan-1-ols* (3b). The yield was 42%, $R_{\rm f}$ 0.28. Found (%): C, 81.67; H, 11.08. $C_{17}H_{28}O$. Calculated (%): C, 82.20; H, 11.36. IR, v/cm⁻¹: 3380, 3050, 3015, 2910, 2840, 1450, 1275, 1085, 1020, 700. ¹H NMR, δ: 0.88 (t, 6 H, Me, J = 5.9 Hz); 1.24—1.47 (m, 14 H, CH₂); 1.65—2.17 (m, 1 H, CH); 4.63 (t, 1 H, C $\underline{\text{H}}$ -Ph, J = 6.3 Hz); 7.21-7.51 (m, 5 H, Ph). ¹³C NMR, δ: 14.15, 14.22 (q, C(10)); 19.68 (q, C(11)); 22.74 (t, C(9)); 27.03 (t, C(6)); 29.69 (t, C(7)); 31.97 (t, C(8)); 32.82 (t, C(3)); 33.08 (d, C(4)); 36.46, 36.78 (t, C(2)); 36.91 (t, C(5)); 75.21 (d, C(1)); 125.94 (d, C(15)); 127.50 (d, C(13)); 128.47 (d, C(14)); 144.99 (s, C(12)). MS, m/z: 230 [M – 18]⁺.

2,6-Dimethyldecan-3-ols* (3c). The yield was 43%, $R_{\rm f}$ 0.50. Found (%): C, 77.05; H, 13.77. $C_{12}H_{26}O$. Calculated (%): C, 77.34; H, 14.07. IR, v/cm⁻¹: 3350, 2940, 2905, 2840, 1455, 1380, 1020, 715. ¹H NMR, δ: 0.83–0.96 (m, 12 H, Me); 1.12–1.58 (m, 13 H, CH₂, CH); 3.24–3.52 (m, 1 H, CH–OH). ¹³C NMR, δ: 13.96 (q, C(10)); 17.52 (q, C(1), C(12)); 19.66 (q, C(11)); 22.61 (t, C(9)); 30.62 (t, C(8)); 33.28 (d, C(6)); 34.09, 34.33 (t, C(5)); 35.14 (t, C(2)); 36.82, 36.99 (t, C(7)); 38.58 (t, C(4)); 73.26 (s, C(3)). MS, m/z: 168 [M – 18]⁺.

9-Methyltridecan-6-ols* (3d). The yield was 42%, $R_{\rm f}$ 0.51. Found (%): C, 77.90; H, 13.82. $C_{14}H_{30}O$. Calculated (%): C, 78.43; H, 14.10. IR, v/cm⁻¹: 3350, 2940, 2905, 2840, 1455, 1380, 1020, 715. ¹H NMR, δ: 0.83–0.96 (m, 9 H, Me); 1.12-1.58 (m, 19 H, CH₂, CH); 3.36-3.59 (m, 1 H, CH–OH). ¹³C NMR, δ: 13.96 (q, C(1), C(13)); 19.55 (q, C(14)); 22.61 (t, C(12)); 23.00 (t, C(2)); 25.34 (t, C(4)); 28.78 (t, C(8)); 29.30 (t, C(11)); 31.97 (t, C(3)); 32.88 (t, C(9)); 33.27 (t, C(7)); 35.29 (t, C(10)); 37.30 (t, C(5)); 71.96 (d, C(6)). MS, m/z: $196 [M - 18]^+$.

5-Ethyl-8-methyltetradecan-5-ols* (3e). The yield was 48%, $R_{\rm f}$ 0.25. Found (%): C, 79.26; H, 13.86. $C_{17}H_{36}O$. Calculated (%): C, 79.61; H, 14.15. IR, v/cm^{-1} : 3370, 2950, 2915, 2850, 1470, 1385, 1110, 1065, 680. ¹H NMR, δ: 0.79–0.91 (m, 12 H, Me); 1.09—1.31 (m, 22 H, CH₂); 1.52—2.01 (m, 1 H, CH); 3.25–3.68 (m, 1 H, C<u>H</u>-OH). ¹³C NMR, δ: 10.84 (q, C(16)); 14.22 (q, C(14)); 14.22 (q, C(1), C(14)); 19.81 (q, C(16)); 22.80 (t, C(2)); 23.45 (t, C(13)); 27.16 (t, C(10)); 29.76 (t, C(11)); 32.04 (t, C(12)); 32.10 (t, C(3)); 33.40 (d, C(8)); 34.83, 35.03 (t, C(15)); 37.11 (t, C(4), C(9)); 39.25 (t, C(7)); 41.66 (t, C(6)); 72.94 (s, C(5)). MS, m/z: 238 [M – 18]⁺.

5-Methyl-2-phenylundecan-2-ols* (3f). The yield was 47%, $R_{\rm f}$ 0.23. Found (%): C, 81.96; H, 11.22. $C_{18}H_{30}O$. Calculated (%): C, 82.38; H, 11.52. IR, v/cm⁻¹: 3380, 3050, 3010, 2920, 2850, 1445, 1375, 1025, 1065, 765. ¹H NMR, δ: 0.68—1.08 (m, 9 H, Me); 1.21–1.47 (m, 14 H, CH₂); 1.60–1.79 (m, 1 H, CH); 7.10—7.71 (m, 5 H, CH—Ph). ¹³C NMR, δ: 14.09 (q, C(11)); 19.61 (q, C(12)); 22.61 (t, C(10)); 27.16 (t, C(7)); 29.56 (t, C(8)); 30.85 (q, C(1)); 31.84 (t, C(9)); 32.95 (t, C(4)); 34.64 (d, C(5)); 36.72 (t, C(6)); 41.46 (t, C(3)); 74.64 (s, C(2)); 124.77 (d, C(14), C(18)); 126.26 (d, C(16)); 127.95 (d, C(15), C(17); 148.11, 148.31 (s, C(13)). MS, m/z: 244 [M – 18]⁺.

6-Ethyldodecan-4-ols* (4a). The yield was 36%, R_f 0.54. Found (%): C, 78.06; H, 13.92. $C_{14}H_{30}O$. Calculated (%): C, 78.43; H, 14.10. IR, v/cm⁻¹: 3380, 2945, 2900, 2850, 1100, 1255, 750. ¹H NMR, δ: 0.86–0.92 (m, 9 H, Me); 1.26–1.58 (m, 19 H, CH₂, CH); 3.48–3.68 (m, 1 H, C<u>H</u>–OH). ¹³C NMR, δ: 10.74 (q, C(14)); 14.06 (q, C(12)); 14.10 (q, C(1)); 22.66 (t, C(11)); 22.83 (t, C(2)); 25.54, 26.35 (t, C(13)); 26.58 (t, C(8)); 29.74, 29.80 (t, C(9)); 31.91 (t, C(10)); 33.69 (t, C(7)); 35.53, 35.32 (d, C(6)); 40.39 (t, C(3)); 41.81, 41.89 (t, C(5)); 69.70 (d, C(4)). MS, m/z: 196 [M – 18]⁺.

3-Ethyl-1-phenylnonan-1-ols* (4b). The yield was 35%, $R_{\rm f}$ 0.36. Found (%): C, 81.96; H, 11.06. $C_{17}H_{28}O$. Calculated (%): C, 82.20; H, 11.36. IR, v/cm⁻¹: 3380, 3050, 3015, 2910, 2840, 1450, 1275, 1085, 1020, 700. ¹H NMR, δ: 0.73—0.84 (m, 6 H, Me); 1.19–1.70 (m, 14 H, CH₂); 2.38–2.51 (m, 1 H, CH); 4.47—4.74 (m, 1 H, CH—OH); 7.13—7.48 (m, 5 H, Ph). ¹³C NMR, δ: 10.84 (q, C(11)); 14.15, 14.22 (q, C(9)); 22.74 (t, C(8)); 25.47 (t, C(10)); 26.96 (t, C(5)); 29.82 (t, C(6)); 31.97 (t, C(7)); 33.40 (t, C(4)); 35.35 (d, C(3)); 43.35, 43.42 (t, C(2)); 72.48 (d, C(1)); 125.87 (d, C(13)); 127.24 (d, C(14)); 128.21 (d, C(15)); 145.38 (s, C(12)). MS, m/z: 230 [M – 18]⁺.

5-Ethyl-2-methylnonan-3-ols* (4c). The yield was 35%, $R_{\rm f}$ 0.57. Found (%): C, 76.96; H, 13.83. $C_{12}H_{26}O$. Calculated (%): C, 77.34; H, 14.07. IR, v/cm⁻¹: 3350, 2940, 2905, 2840, 1455, 1380, 1020, 715. ¹H NMR, δ: 0.83–0.96 (m, 12 H, Me); 1.12-1.58 (m, 13 H, CH₂, CH); 3.24-3.52 (m, 1 H, CH-OH). ¹³C NMR, δ : 10.70 (g, C(12)); 13.96 (q, C(9)); 17.52 (q, C(1), C(10)); 22.61 (t, C(8)); 26.39, 26.52 (t, C(6)); 26.39, 26.52 (q, C(11)); 29.35 (t, C(7)); 34.75, 34.96 (d, C(5)); 35.09 (t, C(2)); 45.31 (t, C(4)); 71.89 (s, C(3)). MS, m/z: 168 [M – 18]⁺.

8-Ethyldodecan-6-ols* (4d). The yield was 35%, $R_{\rm f}$ 0.57. Found (%): C, 78.00; H, 13.81. C₁₄H₃₀O. Calculated (%): C, 78.43; H, 14.11. IR, v/cm⁻¹: 3350, 2940, 2905, 2840, 1455, 1380, 1020, 715. ¹H NMR, δ: 0.83–0.96 (m, 9 H, Me); 1.12-1.58 (m, 19 H, CH₂, CH); 3.36-3.58 (m, 1 H, C<u>H</u>-OH). ¹³C NMR, δ: 10.19, 10.58 (q, C(14)); 13.96 (q, C(1), C(12)); 22.61 (t, C(11)); 23.00 (t, C(2)); 25.34 (t, C(4)); 26.44 (t, C(13)); 29.30 (t, C(10)); 31.97 (t, C(3)); 34.90 (t, C(9)); 36.65, 36.78 (d, C(8)); 38.08 (t, C(5)); 41.66 (t, C(7)); 69.58 (d, C(6)). MS, m/z: 196 [M – 18]⁺.

5,7-Diethyltridecan-5-ols* (4e). The yield was 32%, $R_{\rm f}$ 0.31. Found (%): C, 79.11; H, 13.85. C₁₇H₃₆O. Calculated (%): C, 79.61; H, 14.15. IR, v/cm⁻¹: 3370, 2950, 2915, 2850, 1470, 1385, 1110, 1065, 680. ¹H NMR, δ: 0.76–0.91 (m, 12 H, Me);

^{*} A 1:1 mixture of diastereomers.

1.10-1.32 (m, 22 H, CH₂); 1.52-1.94 (m, 1 H, CH); 3.25-3.65 (m, 1 H, CH-OH). ¹³C NMR, δ : 10.84 (q, C(15), C(17)); 14.22 (q, C(1), C(13)); 22.80 (t, C(2)); 23.45 (t, C(12)); 26.34, 26.70 (t, C(14)); 27.22 (t, C(9)); 29.89 (t, C(10)); 32.04 (t, C(11)); 32.04 (t, C(3)); 32.29 (t, C(16)); 33.01 (t, C(8)); 34.84, 35.03 (d, C(7)); 39.25 (t, C(4)); 45.76 (t, C(6)); 73.52 (s, C(5)). MS, m/z: 238 [M -18]⁺.

4-Ethyl-2-phenyldecan-2-ols* (4f). The yield was 33%, $R_{\rm f}$ 0.29. Found (%): C, 81.92; H, 11.26. $C_{18}H_{30}O$. Calculated (%): C, 82.38; H, 11.52. IR, $v/{\rm cm}^{-1}$: 3380, 3050, 3010, 2920, 2850, 1445, 1375, 1025, 1065, 765. $^{1}{\rm H}$ NMR, δ : 0.76—0.91 (m, 9 H, Me); 1.08—1.40 (m, 14 H, CH₂); 1.62—1.76 (m, 1 H, CH); 7.15—7.51 (m, 5 H, CH—Ph). $^{13}{\rm C}$ NMR, δ : 10.51 (q, C(12)); 14.09 (q, C(10)); 22.61 (t, C(9)); 26.90 (t, C(11)); 26.90 (t, C(6)); 29.56 (t, C(7)); 29.89 (q, C(1)); 31.84 (t, C(8)); 34.12 (d, C(4)); 36.72 (t, C(5)); 48.03 (t, C(3)); 75.67 (s, C(2)); 124.77 (d, C(14), C(18)); 126.26 (d, C(16)); 127.95 (d, C(15), C(17)); 148.11, 148.31 (s, C(13)). MS, m/z: 244 [M — 18]⁺.

CuCl-catalyzed reactions of ACP 5a,b with acetone. The catalyst Cp₂ZrCl₂ (0.5 mmol), hexane (3 mL), endo-dicyclopentadiene (10 mmol), and Et₃Al (12 mmol) were placed in a glass reaction vessel under dry argon at 0 °C. The reactions were carried out according to the general procedure.

2-Methyl-4-(tricyclo[5.2.1.0^{2,6}]dec-4-en-8-yl)butan-2-ol (7a) and 2-methyl-4-(tricyclo[5.2.1.0^{2,6}]dec-3-en-8-yl)butan-**2-ol (7b).** The total yield was 75% (in the ratio $\sim 3:1$), $R_f 0.19$. Found (%): C, 81.35; H, 10.73. C₁₅H₂₄O. Calculated (%): C, 81.76; H, 10.93. IR, v/cm⁻¹: 3395, 3030, 1930, 1450, 1380, 900, 720. ¹H NMR, δ: 0.66–0.95 (m, 6 H, Me); 1.08–2.48 (m, 15 H, CH₂, CH); 5.48–5.66 (m, 2 H, CH=CH). <u>7a</u>: ¹³C NMR, δ: 29.24 (q, C(14), C(15)); 31.58 (t, C(3)); 31.97 (t, C(8)); 34.04 (t, C(11)); 34.08 (d, C(9)); 38.41 (t, C(10)); 40.03 (d, C(7)); 42.37 (t, C(12)); 42.51 (d, C(2)); 46.02 (d, C(1)); 52.32 (d, C(6)); 70.99 (s, C(13)); 130.43 (d, C(4)); 132.57 (d, C(5)). <u>7b</u>: ¹³C NMR, δ: 29.24 (q, C(14), C(15)); 30.67 (t, C(8)); 32.36 (t, C(5)); 34.04 (d, C(9)); 34.08 (t, C(11)); 38.41 (t, C(10)); 41.72 (d, C(7)); 42.37 (t, C(12)); 42.46 (d, C(6)); 44.46 (d, C(1)); 53.69 (d, C(2)); 70.99 (s, C(13)); 130.43 (d, C(4)); 132.57 (d, C(3)). MS, m/z: 212 [M – 18]⁺.

CuCl-catalyzed reactions of ACP 6 with aldehydes. Fullerene[60] (0.025 mmol) in 30 mL of toluene was placed in a glass reaction vessel under dry argon and stirred until C_{60} was completely dissolved. The catalyst Cp_2ZrCl_2 (0.0025 mmol) and Et_3Al (0.75 mmol) were added, and the reaction mixture was stirred for 12 h. Copper(1) chloride (0.01 mmol) was added at -15 °C, and then an aldehyde (0.75 mmol) was slowly added dropwise. The reaction mixture was stirred at ~20 °C for 8 h and treated with a saturated solution of HCl in ether. The products were extracted with toluene, and the extract was chromatographed on Al_2O_3 . The products were isolated by successive elution with toluene and methanol.

1-(3-Hydroxyhexyl)-2-hydrofullerene[60] (8). The yield was 78%. Found (%): C, 92.07; H, 3.96. $C_{60}C_{18}H_{42}O_3$. For n = 3 calculated (%): C, 91.21; H, 4.12. IR, v/cm^{-1} : 3350, 2950, 2880, 1570, 1520, 1460, 1520, 1460, 1380, 1280, 1245, 1090, 1030, 720, 530 (fullerene). ¹H NMR, δ: 0.89 (m, 3 H, Me); 1.21–1.58 (m, 6 H, CH₂); 3.38 (m, 1 H, CH₂OH); 2.22–2.36 (m, 2 H, C_{60} CH₂); 6.40 (s, 1 H, C_{60} CH). ¹³C NMR (CDCl₃,

CS₂), δ : 14.10 (q, C(6)); 21.14 (t, C(5)); 33.22 (t, C(4)); 36.39 (t, C(2)); 37.04 (t, C(1)); 71.73 (d, C(3)); 120.21–157.65 (m, C₆₀).

1-(3-Hydroxy-3-phenylpropyl)-2-hydrofullerene[60] (9). The yield was 70%. Found (%): C, 91.65; H, 3.03. $C_{60}C_{18}H_{42}O_3$. For n=3 calculated (%): C, 92.54; H, 3.21. 1H NMR, δ : 2.22—2.31 (m, 2 H, CH₂); 2.89 (m, 2 H, C₆₀—CH₂); 4.38 (m, 1 H, CH—OH); 6.09 (s, 1 H, C₆₀—H); 7.09—7.59 (m, 5 H, Ph). 13 C NMR (CDCl₃, CS₂), δ : 31.12 (t, C(2)); 41.01 (t, C(1)); 66.24 (d, C(3)); 126.26 (d, C(6), C(8)); 127.76 (d, C(7)); 128.21 (d, C(5), C(9)); 140.57 (s, C(4)); 121.88—156.20 (m, C₆₀).

CuCl-catalyzed reactions of ACP 10 with aldehydes. Tris(n-hexyl)aluminum¹⁸ synthesized from Bu $^{1}_{3}$ Al (12 mmol) and hex-1-ene (36 mmol) was placed in a glass reaction vessel under dry argon at 0 °C. Then, hex-1-ene (10 mmol) and Cp₂ZrCl₂ (0.5 mmol) were added, and the reaction mixture was stirred at ~20 °C for 12 h. The reactions were carried out according to the general procedure.

6-Butyl-7-methylundecan-4-ols* (11a). The yield was 78%, $R_{\rm f}$ 0.36. Found (%): C, 79.02; H, 13.88. $C_{16}H_{34}O$. Calculated (%): C, 79.26; H, 14.14. IR, v/cm^{-1} : 3385, 2950, 2915, 2865, 2840, 1475, 1385, 1155, 1020, 910, 720, 690. ¹H NMR, δ : 0.72—1.02 (m, 12 H, Me); 1.21—1.53 (m, 18 H, CH₂); 1.76—2.15 (m, 2 H, CH); 3.48—4.01 (m, 1 H, CH—OH). ¹³C NMR, δ : 14.02 (q, C(1), C(11), C(15)); 18.77, 18.96 (q, C(16)); 22.61 (t, C(2)); 23.00 (t, C(10), C(14)); 29.89 (t, C(9)); 30.02, 30.15 (t, C(12)); 31.71, 31.85 (t, C(13)); 33.01, 33.73 (t, C(8)); 36.13, 36.33 (t, C(3)); 37.50, 38.02 (d, C(6)); 38.47, 38.99 (d, C(7)); 39.77, 40.62 (t, C(5)); 69.49, 70.14 (d, C(4)). MS, m/z: 224 [M — 18]⁺.

5-Butyl-2,6-dimethyldecan-3-ols* (11b). The yield was 76%, $R_{\rm f}$ 0.33. Found (%): C, 78.89; H, 13.86. $C_{16}H_{34}O$. Calculated (%): C, 79.26; H, 14.14. IR, v/cm^{-1} : 3380, 2950, 2920, 2870, 2845, 1480, 1385, 1150, 1020, 900, 720, 680. ¹H NMR, δ: 0.72—0.94 (m, 15 H, Me); 1.12—1.47 (m, 14 H, CH₂); 1.87—2.11 (m, 2 H, CH); 3.28—3.53 (m, 1 H, CH—OH). ¹³C NMR, δ: 14.02 (q, C(10), C(14)); 16.49, 17.27 (q, C(1), C(15)); 18.83, 19.03 (q, C(16)); 23.00 (t, C(9), C(13)); 29.89 (t, C(8)); 30.08 (t, C(12)); 31.12 (t, C(11)); 33.34, 33.92 (t, C(7)); 34.05, 34.25 (d, C(2)); 34.51, 35.09 (d, C(5)); 38.15 (d, C(6)); 39.45 (t, C(4)); 74.43, 75.02 (d, C(3)). MS, m/z: 224 [M — 18]⁺.

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^{*} A 1:1 mixture of diastereomers.

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